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## Direct Resolution of Secondary *Tert*-Butylphenylphosphine Oxide

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# Direct Resolution of Secondary Tert-Butylphenylphosphine Oxide

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Secondary phosphine oxides constitute an important class of organophosphorus compounds and are higly useful as versatile phosphinoylating agents and as ligands [11.2]. They are however not readily available in the optically active form. [13.5] We wish to demonstrate that P-chiral secondary tert-butylphenylphosphine oxide (1) can be promptly resolved into enantiomers by means of its diastereoisomeric complexes with L-O, O- dibenzoyltartaric acid (L-DBTA) Dissolution of equimolar amounts of racemic 1 and L-DBTA in the 4-1 benzene acetone mixture yields crystalline 1.1 complex containing exclusively the R-enantiomer of the starting phosphine oxide. The complexed S-enantiomer is obtained from the mother liquor by crystallization from benzene. The enantiomers of 1 are freed from their L-DBTA complexes by simple washing with aqueous NaOH Both enantiomers of 1 of very high optical purity are thus obtained from a single batch. These enantiomers have subsequently been tested for their configurational stability in selected synthetic applications

The two diastereoisomeric 1 · 1 complexes have been studied by a single-crystal X-ray diffraction method and their secondary structure is visualized above. It follows form this study that P-H hydrogens are not involved in the association process.

#### References

- R.Engel, Synthesis of Carbon-Phosphorus Bonds (CRC Press, Inc.: Boca Raton FL, 1988).
- [2] A.W. Parkins, Platinum Metals Rev., 40, 4, 169 (1996).
- [3] T.L. Emmick, R.L. Letsinger, J. Am. Chem. Soc., 90, 3459 (1968).
- [4] J. Michalski, Z. Skrzypczynski, J. Org. Chem., 97 (1975).
- [5] R. Freeman, R.K. Haynes, W.A. Loughlin, C. Mitchell, J.V. Stockes, Pure & Appl. Chem 65, 647 (1993).